

## Effect of anthocyanin to dissolve A $\beta$ amyloid fibrils studied by TIRFM-QCM biosensor.

TIRFM-QCM を用いた A $\beta$  アミロイド線維に対する  
アントシアニンの融解効果の研究

Kohei Kamada<sup>1†</sup>, Kentaro Noi<sup>1</sup>, and Hirotsugu Ogi<sup>1</sup> (<sup>1</sup>Grad. School Eng., Osaka Univ.)  
鎌田 航平<sup>1†</sup>, 野井 健太郎<sup>1</sup>, 荻 博次<sup>1</sup> (大阪大院・工)

### 1. Introduction

Alzheimer's disease (AD) is the most common dementia. In 1907, Alois Alzheimer described pathological alterations in the brain of a patient of dementia which was named Alzheimer's disease. Despite the fact that intensive efforts have been made by many researches, there is still no effective method of treatment for AD. Many scientists and medical doctors recognize that deposition of amyloid  $\beta$  (A $\beta$ ) peptides and their aggregation on neurons cause AD. A $\beta$  takes several phases, including monomer, oligomer, and fibril. Especially, A $\beta$  amyloid fibril is always confirmed at AD patients' neuron. As a candidate method for AD treatment, some researches focus on dissolving A $\beta$  amyloid fibril. It has reported that some of anthocyanins (a kind of polyphenol) can inhibit the fibril formation in the bulk solution.<sup>1</sup> However, there is no such a report on a solid surface, closer to surroundings in vitro, and there is no systematic study on anthocyanin's ability to dissolve A $\beta$  amyloid fibril. To monitor formation of A $\beta$  amyloid fibrils and their dissolving process by anthocyanin on the solid surface, we use the TIRFM-QCM assay system. It is the combination of the total internal reflection fluorescence microscopy (TIRFM) to monitor the change in the fibril structure and wireless-electrodeless quartz crystal microbalance (WE-QCM) biosensor to monitor the change of amount of A $\beta$  molecules on the surface. Here, we use delphinidin 3-galactoside (DP3G: a kind of anthocyanin) for dissolving A $\beta$  amyloid fibrils.

### 2. TIRFM-QCM Biosensor

**Fig.1** shows illustrates the TIRFM-QCM biosensor measurement. The WE-QCM device is set on the objective lens of TIRFM. The flow channel is set on a cover glass, along which the A $\beta$ <sub>1-40</sub> solution or the anthocyanin solution flow. The quartz oscillator is put in the middle of the flow

channel, whose vibration is excited by the antenna for transmission by applying alternating electric field, and the vibrational signals are detected by the other antenna through the piezoelectric effect.

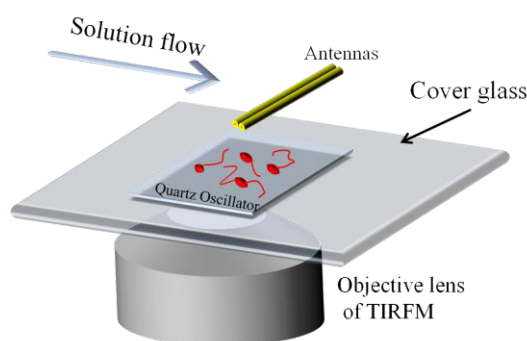


Fig.1 Schematic illustration of TIRFM-QCM biosensor.

TIRFM is a type of fluorescence microscopes. The incident light is totally reflected on the surface of quartz oscillator and then the evanescent light is generated. TIRFM uses the evanescent light as the excitation light, so that the fluorescence images become dramatically clear because of the absent of the scattered excitation lights. It is effective to observe the change in the fibril structure by adding thioflavin T (ThT) molecules in the flowing solution because ThT emits strong light when it binds with the  $\beta$ -sheet structure of fibrils. Because biosensor chips generally have metal coatings on their surfaces, it is difficult to perform the TIRFM observation on such chips. The WE-QCM chip has no coating, thus, we can use the WE-QCM for TIRFM. This is the first assay system which achieves combination of QCM and TIRFM. Furthermore, it has high mass sensitivity because the sensitivity of the QCM deteriorates with the heavy metallic coatings.

### 3. Experimental Details

#### 3.1 Monitoring formation of A $\beta$ amyloid fibril

We irradiated the A $\beta$ <sub>1-42</sub> amyloid-fibril solution with ultrasound to break the fibrils down into fragments. They were fixed on the quartz oscillator as seeds for A $\beta$  amyloid fibril. After the

resonant frequency of WE-QCM became sufficiently stable while flowing buffer solution (ultra pure water with 30  $\mu\text{M}$  ThT), we let the water solution including 10  $\mu\text{M}$   $\text{A}\beta_{1-40}$  peptides and 30  $\mu\text{M}$  ThT flow over the seeded quartz oscillator and monitored formation of  $\text{A}\beta$  fibril for 10 h.

### 3.2 Monitoring $\text{A}\beta$ fibril Dissolving process

After forming the  $\text{A}\beta_{1-40}$  fibrils on the quartz surface, we let the water solution including 80  $\mu\text{M}$  DP3G and 30  $\mu\text{M}$  ThT flow over the quartz oscillator with the fibril and monitored the dissolving process of  $\text{A}\beta$  amyloid fibril for 10 h.

For a control experiment, we let the water solution including 80  $\mu\text{M}$  chlorogenic acid hemihydrate (a type of polyphenols) and 30  $\mu\text{M}$  ThT flow over the oscillator for 10 h.

## 4. Results and Discussion

**Fig.2** shows the fluorescent images by TIRFM during formation of  $\text{A}\beta$  amyloid fibrils. A few hours later, the  $\text{A}\beta$  amyloid fibril emerged on the surface of the oscillator. We thus succeed in forming  $\text{A}\beta$  amyloid fibrils, which agrees with our previous study<sup>2</sup>.

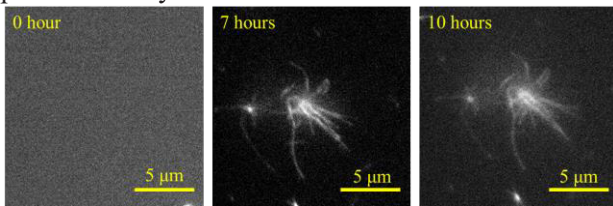


Fig.2 The fluorescence images by TIRFM in forming  $\text{A}\beta$  amyloid fibril. Left shows the image when the flow  $\text{A}\beta_{1-40}$  solution is injected (we set this time to be 0 h). Middle shows an image when the amyloid fibril appeared. Right shows the image at the end of the measurement.

**Fig.3** shows the resonance frequency change measured by the WE-QCM during the flow of DP3G, and **Fig. 4** shows the corresponding fluorescent images taken by TIRFM. From Fig.3, the frequency increase indicates the decrease of mass on the surface of the quartz oscillator. The increase of about 30 ppm means the decrease of about 2 pmol of  $\text{A}\beta_{1-40}$  monomers from the surface. From Fig.4, the light lines become weak and then disappear. Therefore, DP3G seems to have ability to dissolving  $\text{A}\beta$  amyloid fibril.

**Fig.5** shows the fluorescent images by the TIRFM of the control experiment by the chlorogenic acid hemihydrate, where we clearly see the  $\text{A}\beta$  amyloid fibril even after the flow of the control anthocyanin for 10 h. These images significantly differ from those in the DP3G experiment. Thus, it is obvious that DP3G has a strong effect to dissolve  $\text{A}\beta$  amyloid fibril.

Therefore, we succeeded to analyze the forming and dissolving  $\text{A}\beta$  amyloid fibril from two different viewpoints, mass changes and fluorescence imaging, by TIRFM-QCM. We will make experiments for other anthocyanins to evaluate their  $\text{A}\beta$ -fibril dissolving ability, and investigate its correlation with the number and position of hydroxyl group in the anthocyanin molecule.

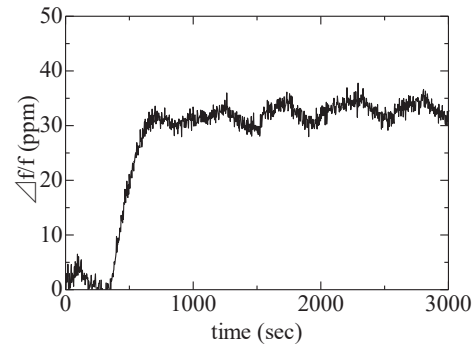


Fig. 3 The resonance frequency change by WE-QCM of dissolving  $\text{A}\beta$  amyloid fibril by DP3G. Time=0 means the time when started to flow DP3G solution.

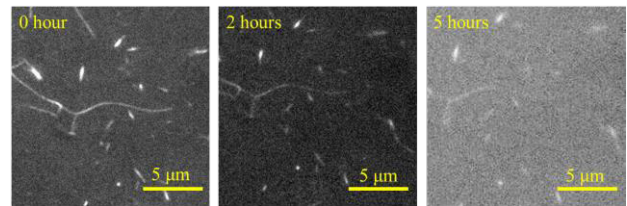


Fig.4 The fluorescence images by TIRFM in dissolving  $\text{A}\beta$  amyloid fibril. The left image is the image when time of starting to flow DP3G (0 h). The middle one is the image when started dissolving the amyloid fibril. The right one is the image 5 h later. Then, fibril almost disappeared.

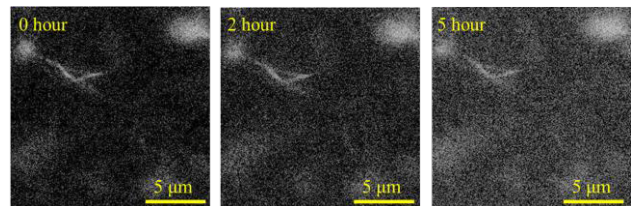


Fig.5 The fluorescence images by TIRFM in a control experimental. 5 h later, the fibril had almost no change.

## References

1. H. Iwasa, H. Kameda, N. Fukui, S. Yoshida, K. Hongo, T. Mizobata, S. Kobayashi and Y. Kawata: *Biochemistry* **52** (2013) 9202.
2. H. Hamada, H. Ogi, K. Noi, H. Yagi, Y. Goto, and M. Hirao: *Jpn. J. Appl. Phys.* **54** (2015) 07HE01.